ABSTRACT. A shallow firn core, recovered from the southern Patagonia ice cap (elevation 2680 m a.s.l.), has been analyzed for H stable isotope composition and for major soluble chemical compounds. The temperature measurement at 13.17 m depth (bottom of the core) shows that the ice cap is temperate. The chemical profiles indicate that some soluble impurities have been partly washed out by percolation, but seasonal deuterium content variations are relatively well preserved down to the bottom of the core, which allows the net accumulation rate of the site to be calculated (1.2 m water equivalent). The mean concentrations measured in the most recent year recorded (Cl⁻: 4.85, nssSO₄²⁻: 0.65, NO₃⁻: 0.32, Na⁺: 4.30, K⁺: 0.64 and NH₄⁺: 0.77, in μeq.l⁻¹) serve as a reference for the background chemical composition of precipitation at mid-southern latitudes under South Pacific meteorological conditions.

INTRODUCTION

Snow accumulated in polar areas or on high mountains records environmental parameters, such as temperature and atmospheric composition, which are of great interest for studying the temporal evolution of climate and the composition of the remote atmosphere. Ideal conditions are found in central Antarctic regions where the very low temperatures and accumulation rates enable the formation of reliable atmospheric records spanning several hundred millennia. On the other hand, the meteorological and geographical conditions are generally less favorable at mid or low latitudes: ice-covered areas are much smaller, ice-flow patterns are frequently intricate (which makes ice-core dating uncertain) and overall higher temperatures are the cause of a more or less pronounced melting phenomenon which may perturb significantly the information contained in the deposited snow layers. Nevertheless, two of the longest non-polar isotope and chemical parameter records were obtained from temperate glaciers: Quelccaya Ice Cap (Thompson and others, 1988) and Vernagtferner (Baker and others, 1985). These limitations, in addition to the practical difficulties of working at very high altitudes, may explain why glacio-chemical data from high mountainous areas of the world are scarce, even though such research is strongly encouraged by scientific committees (e.g. SCAR (1974), for the Patagonian ice caps) dealing with regional or global climatic and environmental problems. The environmental data available for mid- and low-latitude glaciers have been reviewed recently by Wagenbach (1989).

In spite of the fact that the Patagonian ice caps are the most extensive non-polar ice caps (18,000 km²; Fig. 1), they have been poorly studied in the past, in particular regarding glaciology and atmospheric chemistry. Little is known of the snow-accumulation rates in the area, the ice-cap thickness, the temperature or the movement of the ice. The chemical composition of the local precipitation has never been determined. The only relevant investigations presently published are those of Nakajima (1985) and Japanese Society of Snow and Ice (1987) on the northern Patagonia ice cap. A 37.5 m firn core was drilled in 1985 (Yamada, 1987), but unfortunately at a site where high melting and frequent rain events prevent any use of the ice samples for further isotopic and chemical investigations.

The purpose of this preliminary study is to document some of the parameters listed above in order to evaluate the feasibility of a more complete investigation of the area in the future, primarily based on a deep ice core which could be drilled on the southern Patagonia ice cap (SPIC), which is considered to be more suitable for such studies than the northern ice cap.

We report in this paper the analysis of a 13.17 m deep firn core drilled on the SPIC on 9 September 1986. This is the most recent ice core available in this region.

SAMPLING CONDITIONS AND LABORATORY ANALYTICAL METHODS

Field work

The SPIC, about 350 km long and 30-40 km wide, is situated between 48°20' and 51°30'S and 73° and 74°W.
The drilling site (50°38'S, 73°15'W, 2680 m a.s.l.) was located on a 100 km² gently sloping plateau, which is the accumulation area for several large glaciers (Fig. 1). The sampling equipment, in particular the hand auger, and the operators were transported by helicopter from the edge of one of these glaciers (Perito Moreno). The firn core, placed immediately in an insulated box after drilling, was rapidly transferred to a freezer installed in a tourist bungalow located in the vicinity of the glacier. The temperature was measured at the bottom of the borehole 48 h after recovery of the firn core with the aid of a calibrated thermistor.

The firn core was cut into 45 sections at the bungalow. After density measurements, the core sections were subsampled for further isotope determinations. For chemical studies, the external part of each sample was removed using pre-cleaned tools by operators wearing clean-room clothes and masks. The firn samples were then transferred into sealed polyethylene bags and kept frozen until their analysis in Grenoble.

**Laboratory work**

The samples were analyzed in the laboratory in clean-air conditions, according to the procedure of Legrand and Delmas (1987). Major ions were determined by ion chromatography (Dionex 2010). The estimated precision was 10%. The stable-isotope content (δD‰) was measured by mass spectrometry on the sub-samples collected in the field at Centre d’Études Nucléaires de Saclay, France.

**RESULTS AND DISCUSSION**

**Temperature**

The temperature measured at the bottom of the borehole (13.17 m) was exactly 0°C, indicating a temperate glacier. The mean annual temperature of the area, even at lower altitudes, is very poorly documented. We estimated the mean annual temperature at the drilling site tentatively using data collected at the meteorological stations of Lago Argentino (50°20'S, 72°18'W, 200 m a.s.l.), about 80 km to the northeast, and Torres del Paine (51°10'S, 72°18'W, 125 m a.s.l.), about 60 km to the southwest. The mean annual temperatures there were 7.3°C (Servicio Meteorológico Nacional de la República Argentina) and 7°C (personal communication from W.C. Keene), respectively. The altitudinal temperature lapse rate used in the calculation is 0.53°C/100 m. This value, which is very close to the net adiabatic lapse rate, was obtained by Inoue and others (1987) for the northern Patagonia ice cap. The temperature calculated in this way is approximately −6°C, a figure relatively low in comparison with the Quelccaya Ice Cap (−3°C), which prevents excessive percolation.

**Density**

The physical appearance of the shallow snow layers is typical of a temperate glacier, with coarse grains and numerous ice layers 0.5-30 mm thick. Consequently, the density (d) profile is irregular, which is also characteristic of temperate glaciers. It can be represented by the linear regression ($R^2 = 0.78$) of the 37 levels measured: $d = 20z + 406$ (d in kg m⁻³ and z in m). The mean value (606 kg m⁻³) at 10 m depth and the transition depth of firn to ice (20-25 m, by extrapolation) are comparable to corresponding values obtained on the Quelccaya Ice Cap (540 kg m⁻³; Thompson, 1980) and on Vernagtferner (640 kg m⁻³ and 20-25 m; Oerter and others, 1983). Similar values were also obtained on the northern Patagonia ice cap (Yamada, 1987), despite the different densification process due to the large amount of water present in the snow.

**Isotope measurements, dating and snow-accumulation rate**

The isotopic composition (O and H atoms) of polar snow depends on atmospheric temperature (Dansgaard, 1964). Temperature changes, in particular seasonal cycles, are therefore naturally recorded in polar ice caps. These seasonal variations are commonly used for dating firn and ice layers (Hammer, 1989). On the other hand, for glaciers located at mid or tropical latitudes, meltwater percolation tends to destroy seasonal patterns and to homogenize the isotopic composition. For temperate glaciers, the effect is particularly well marked, and only in special cases (conjunction of particular meteorological and physical conditions) do the isotopic profiles still
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Fig. 2. Seasonal variations of the deuterium content compared with the snow stratigraphy. Water-equivalent depth (7.07 m at the bottom) is calculated using the mean density profile. Black segments represent ice layers (0.5–30 mm thick). Open circles represent ice lenses.

contain information usable for climatic and dating purposes (Stichler and others, 1983; Thompson and others, 1984). At our site, the relatively low estimated mean annual temperature (-6°C) and local meteorological conditions (high precipitation rate, prevailing westerly Pacific winds and frequent cloudiness which reduced summer melting) are favourable factors for the preservation of the initial stratification of the snow layers. The isotopic profile (Fig. 2) exhibits fluctuations which can reasonably be attributed to seasonal variations. The core was dated on the basis of the visual glaciological stratigraphy reported in Figure 3, combined with the interpretation of this isotope profile. The winters of 1982, 1984, 1985 and 1986 (date of field work) and the summers of 1980–81, 1983–84, 1984–85 and 1985–86 are clearly identified.

A mean value of 1.2 m water equivalent (w.e.) is deduced for the annual accumulation rate. This is only one-third of what can be expected from data obtained at nearby temporary stations (Agua y Energía, 1978) using their precipitation versus longitude gradient. Such a difference, also observed on the northern Patagonia ice cap (Fujiyoshi and others, 1987; Yamada, 1987), could be caused by the strong and frequent winds prevailing in the area (which blow away snow layers) and/or by some surface melting. This second explanation is less likely because of the absence of significant percolation patterns, as mentioned earlier. Finally, we cannot entirely rule out an erroneous interpretation of the annual signals, which could lead to a smaller value for the number of years encompassed by this core due, for instance, to the occurrence of two isotope peaks per year as is sometimes observed (for example, at Mont Blanc by Jouzel and others (1984)).

Chemical measurements

In temperate glaciers, the depth profiles of chemical species are generally significantly disturbed by meltwater percolation. When a water table forms, most of the soluble impurities are washed out and seasonal variations may be replaced by other fluctuations linked with the up and down movement of the water level. No information is available on whether this phenomenon occurs in the deeper firn layers. In the upper firn layers, a memory of the initial seasonal variations can persist for several years, provided percolation is not too intense. This seems to be the case at the study site, according to visual observations in the field (the ice core was "dry" and the borehole free of water) and to the persistence of the seasonal isotopic patterns.

Chemical depth profiles are shown in Figure 3. Sulfate and nitrate concentrations both exhibit a rapid decrease (by more than 50%) in the first 3 m (i.e. about 2 years). This part of the profile can be represented satisfactorily by exponential functions with $r^2 = 0.68$ and 0.66 for SO$_4^{2-}$ and NO$_3^-$, respectively. The ratio Cl$^-$/Na$^+$ is 1.13 for the first year of accumulation, a value very close to the marine reference value (1.16). This observation is in agreement with the marine origin of the air masses in the area, as was also found at Torres del Paine (Table 1; Cl$^-$/Na$^+$ = 1.17). However, this ratio decreases markedly as a function of depth (Fig. 3), which indicates that Cl$^-$ is preferentially eluted to Na$^+$. Below a depth of 3 m, nitrate values continue to decrease, but more slowly, whereas sulphate values remain relatively stable around 0.4 µeq. l$^{-1}$. For Cl$^-$ and Na$^+$, the decrease is much less marked and high values are still observed after 4 or 5 years. The K$^+$ and NH$_4^+$ profiles appear to be independent of depth.

We can therefore conclude that the following elution sequence is obtained:

in agreement with the results of Davies and others (1982), which concerned a site where percolation was so high that the chemical impurities were removed after a single year. In our case, as indicated by the mean value of the Cl$^-$/Na$^+$ ratio, little loss of solute occurred in the upper part
For NH₄⁺, are lower at our southern Patagonia site than concentrations of most elements are markedly disturbed only lower bounds. These data are compared to the levels of chemical impurities in the local precipitation, records. year of precipitation (six samples, from the winter of 1985 by percolation and cannot be used as reliable atmospheric bearing in mind, however, that these values are possibly higher on the southern Patagonia ice cap, in comparison with what might be expected from the marine source. Nevertheless, potassium concentrations are relatively high on the southern Patagonia ice cap, in comparison with the marine reference ratio (0.15) reported in Table 1 is significantly higher than both the marine reference ratio (0.023) and the ratio reported for Torres del Paine precipitation (0.07). Excess potassium could be linked to either a local crustal source or a contribution by volcanic ash may well be preserved satisfactorily in deeper ice cores recovered from the Quelccaya Ice Cap (about 1.5 x 10³ year; Thompson and others, 1986, 1988), where a useful conductometric profile was obtained, and to northern Spitsbergen, where pH and conductivity measurements cover possibly 6000 years (Fujii and others, 1990). Punning and others (1986) claimed they observed seasonal Cl⁻ variations at considerable depths in Nordaustlandet, Svalbard.

**CONCLUSION**

The most important physical and chemical parameters relevant to past climate and environmental studies have been determined for the upper part of the temperate southern Patagonia ice cap. Although ice melting prevents the preservation of the initial atmospheric parameters below the surface layers, δD seasonal variations, preserved down to 10 m depth, combined with visual stratigraphy, allow the dating of the first five annual snow layers. Most measured chemical species exhibit a sharp decrease in concentration after about 1 year. The mean concentrations obtained from the last year of accumulation (winter 1985 to winter 1986) may serve as a reference for the chemical composition of precipitation at these latitudes, where South Pacific meteorological conditions prevail.

Such glaciological characteristics are apparently inadequate for long-term palaeoenvironmental studies. Nevertheless, we can reasonably expect that insoluble atmospheric impurities like crustal dust, soot particles or volcanic ash may well be preserved satisfactorily in deeper firm and ice layers (as demonstrated convincingly by the deep ice-core study reported by Thompson and others (1986, 1988) for the Quelccaya Ice Cap), making this ice cap a potential site for obtaining records of these important atmospheric tracers over several centuries. Little is known about the environmental history of these latitudes on such a time-scale. A deep ice core drilled at this location would be most valuable for documenting the
Table 1. Chemical concentrations (C, μeq. l⁻¹) and deposition fluxes (Φ, μM m⁻² a⁻¹) in snow at the study site (southern Patagonia ice cap). The values are mean values obtained from the six shallowest samples (about 1 year of precipitation, see text). Precipitation chemistry data from Torres del Paine precipitation station (51°10' S, 72°58' W, 125 m a.s.l.; Likens and others, 1987) and from James Ross Island, Antarctic Peninsula (mean values over 130 years; Aristarain and others, unpublished), are also reported for comparison. Corresponding deposition fluxes (Φ) are calculated using the water-deposition rates \( R_w \) (m of water a⁻¹).

<table>
<thead>
<tr>
<th></th>
<th>( R_w )</th>
<th>Cl⁻</th>
<th>nssSO₄²⁻ (1)</th>
<th>NO₃⁻</th>
<th>Na⁺</th>
<th>K⁺</th>
<th>NH₄⁺ (2)</th>
<th>Cl⁻/Na⁺</th>
<th>K⁺/Na⁺</th>
</tr>
</thead>
<tbody>
<tr>
<td>Southern Patagonia</td>
<td>1.20</td>
<td>C</td>
<td>4.85</td>
<td>0.65</td>
<td>0.32</td>
<td>4.30</td>
<td>0.64</td>
<td>0.77</td>
<td>1.13</td>
</tr>
<tr>
<td>ice cap (SPIC)</td>
<td>(3)</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Torres del Paine (TP)</td>
<td>0.75</td>
<td>C</td>
<td>21.9</td>
<td>2.1</td>
<td>0.5</td>
<td>18.7</td>
<td>1.4</td>
<td>0.7</td>
<td>1.17</td>
</tr>
<tr>
<td></td>
<td>(4)</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
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<td></td>
</tr>
<tr>
<td>James Ross Island</td>
<td>0.59</td>
<td>C</td>
<td>7.17</td>
<td>0.68</td>
<td>ND</td>
<td>ND</td>
<td>ND</td>
<td>ND</td>
<td>ND</td>
</tr>
<tr>
<td>(Antarctic Peninsula)</td>
<td>(5)</td>
<td></td>
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<td></td>
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</tbody>
</table>

(1) nssSO₄²⁻ (non-sea-salt sulphate) was calculated as follows from measured sulphate (SO₄²⁻): \[ \text{nssSO}_4^{2-} = [\text{SO}_4^{2-}] - 0.104[\text{Cl}^-] \], where 0.104 is the reference value of the ratio \([\text{SO}_4^{2-}]/[\text{Cl}^-]\) in bulk sea water.

(2) For ammonium, the value of the shallowest sample was excluded because of contamination.

(3) This work.

(4) W. C. Keene (personal communication).


ND Not determined.

past (e.g. climate fluctuations, volcanic eruptions) of this remote part of the Southern Hemisphere.

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